

p-Tolyl 2-*O*-benzyl-3,5-*O*-(di-*tert*-butylsilanediyl)-1-thio- α -D-arabinofuranosideRuel C. Nacario,^a Todd L. Lowary^a and Robert McDonald^{b*}^aAlberta Ingenuity Centre for Carbohydrate Science, Chemistry Department, University of Alberta, Edmonton, Alberta T6G 2G2, Canada, and ^bX-ray Crystallography Laboratory, Chemistry Department, University of Alberta, Edmonton, Alberta T6G 2G2, CanadaCorrespondence e-mail:
bob.mcdonald@ualberta.ca

Key indicators

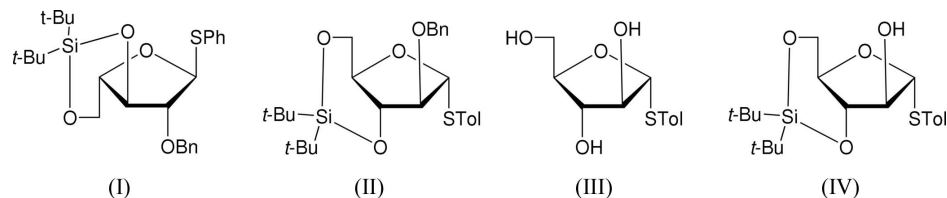
Single-crystal X-ray study
T = 193 K
Mean σ (C–C) = 0.003 Å
R factor = 0.036
wR factor = 0.091
Data-to-parameter ratio = 18.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, C₂₇H₃₈O₄SSi, the furanose ring adopts a nearly perfect *E*₄ envelope conformation, while the six-membered ring containing the Si atom exists in an approximate half-chair conformation. The conformation about the glycosidic linkage is that favored by the *exo*-anomeric effect.

Received 14 December 2006
Accepted 16 December 2006

Comment

A recent paper by Zhu *et al.* (2006) reports an efficient and highly stereoselective synthesis of β -L-arabinofuranosides using a thioglycoside donor protected at O3 and O5 with a silyl acetal (I). Our group has a long-standing interest in the synthesis of D-arabinofuranosides from mycobacteria (Yin *et al.*, 2002; Yin & Lowary, 2001; Han *et al.*, 2003) and thus we prepared an analogous donor, (II), from the previously reported D-arabinofuranosyl thioglycoside, (III) (D'Souza *et al.*, 2000), in two steps *via* (IV) in 88% overall yield. The title compound, (II), was crystalline and an X-ray study was undertaken to confirm the structure of the molecule and also to make structural comparisons with computational work reported earlier on (I) (Zhu *et al.*, 2006).



The attachment of the silicon acetal to the monosaccharide results in the furanose ring being locked into a near-perfect *E*₄ conformation in which C4 is displaced below the plane formed by C1, C2, C3 and O4 (Fig. 1). The pseudorotational phase angle (*P*) of the furanose ring is 57.0° and the puckering amplitude (τ_m) is 43.1° (Altona & Sundaralingam, 1972). This conformation differs slightly from previous density functional theory calculations on (I), which suggested an envelope conformation with C3 displaced from the plane (Zhu *et al.*, 2006). The six-membered ring in (II) exists in a distorted half-chair conformation, in which Si, O3, O5, and C5 are approximately coplanar while C4 is displaced below, and C3 above, this plane. Here too there are slight differences from the computed structure of (I) in which the conformation of this ring was shown to be more chair-like (Zhu *et al.*, 2006). The conformation about the C1–S bond in (II) is that favored by the *exo*-anomeric effect (Lemieux & Koto, 1974), the C11–S–C1–C2 torsion angle being –164.11 (13)°. In the *E*₄ conformation adopted by the furanose ring, the thioglycoside

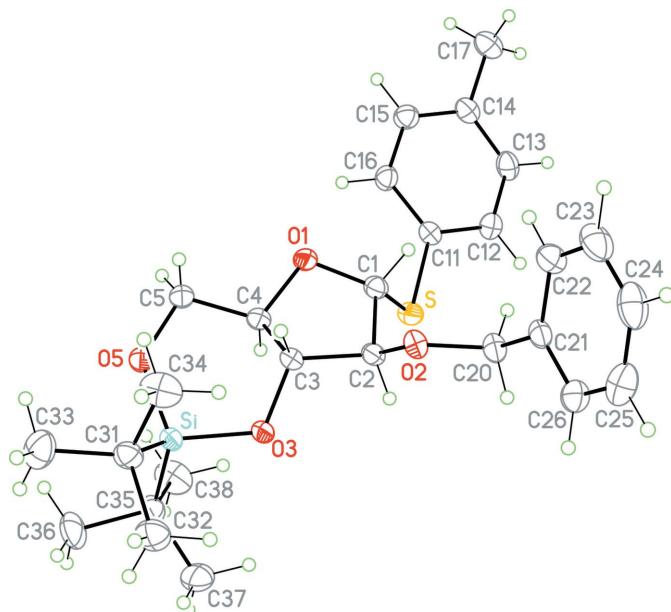


Figure 1
The molecular structure of (II). Displacement ellipsoids are drawn at the 30% probability level. H atoms are shown as arbitrarily small spheres.

unit is oriented pseudoaxially and the benzyloxy substituent at C2 is pseudo-equatorially disposed.

Experimental

p-Tolyl 3,5-*O*-(di-*t*-butylsilylanediyl)-1-thio- α -D-arabinofuranoside (IV). To a solution of (III) (3.00 g, 11.7 mmol) in CH_2Cl_2 (92 ml) and DMF (19 ml) at 273 K was added 2,6-lutidine (5.50 ml, 47.2 mmol) and di-*t*-butylsilyl bis(trifluoromethanesulfonate) (3.93 ml, 10.8 mmol). The reaction mixture was stirred for 4.5 h, after which it was concentrated *in vacuo*, diluted with EtOAc and washed successively with water and brine. The organic layer was dried (MgSO_4), filtered and concentrated *in vacuo* to give a residue that was purified by flash column chromatography (hexanes, 10:1 hexanes/EtOAc) to afford (IV) as a white, amorphous solid (yield 3.17 g, 93%). Data for (IV): R_F 0.49 (5:1 hexanes/EtOAc); $[\alpha]_D^{25} +155.6$ (c 0.7, CH_2Cl_2); ^1H NMR (500 MHz, CDCl_3): δ 7.42 (d, 2H, $J = 8.1$ Hz, ArH), 7.13 (d, 2H, $J = 8.4$ Hz, ArH), 5.25 (d, 1H, $J = 5.9$ Hz, H-1), 4.31–4.34 (dd, 1H, $J = 4.6, 4.6$ Hz, H-5), 4.12–4.15 (m, 1H, H-2), 3.98–4.02 (m, 1H, H-3), 3.92–3.96 (m, 1H, H-5'), 3.88–3.91 (m, 1H, H-4), 2.33 (s, 3H, ArCH₃), 2.57 (d, 1H, $J = 3.2$ Hz, OH), 1.06 [s, 9H, C(CH₃)₃], 0.98 [s, 9H, C(CH₃)₃]; HRMS (ESI) m/z calculated for $\text{C}_{20}\text{H}_{32}\text{O}_4\text{SSi}$ + Na: 419.1683, found: 419.1680. Analysis calculated for $\text{C}_{20}\text{H}_{32}\text{O}_4\text{SSi}$: C 60.57, H 8.13, S 8.08%; found: C 60.09, H 8.06, S 7.66%.

p-Tolyl 2-*O*-benzyl-3,5-*O*-(di-*t*-butylsilylanediyl)-1-thio- α -D-arabinofuranoside (II). Benzyl bromide (1.66 ml, 14.0 mmol) and NaH (0.36 g, 15.1 mmol) were added to a solution of (IV) (1.50 g, 3.78 mmol) in dry THF (28 ml) and the reaction mixture was stirred for 9.5 h at 273 K. The reaction was quenched by the addition of CH_3OH and the reaction mixture was concentrated *in vacuo*. The residue was diluted with CH_2Cl_2 and washed sequentially with a solution of 1 N HCl, a saturated solution of NaHCO_3 , water and brine. The organic phase was dried (MgSO_4), filtered and concentrated *in vacuo*. The residue was then purified by flash column chromatography (hexanes, 15:1 hexanes/EtOAc) to give the product as a white, amorphous solid (yield 1.75 g, 95%), which was recrystallized from hexanes–EtOAc (m.p. 349–351 K). Data for (II): R_F 0.77 (5:1 hexanes/EtOAc); $[\alpha]_D^{25} +157.2$ (c 0.8, CH_2Cl_2); ^1H NMR (600 MHz, CDCl_3): δ 7.41–7.43 (m, 2H, ArH), 7.31–7.39 (m, 5H, ArH), 7.09–7.10 (m, 2H, ArH), 5.36 (d, 1H, $J = 5.4$ Hz, H-1), 4.80 (ABq, 2H, $J = 12.0$ Hz, PhCH₂), 4.32–4.34 (m, 1H, H-5), 4.13–4.15 (m, 1H, H-3), 3.97–3.98 (m, 1H, H-2), 3.95–3.96 (m, 1H, H-5'), 3.89–3.92 (m, 1H, H-4), 2.33 (s, 3H, ArCH₃), 1.08 [s, 9H, C(CH₃)₃], 0.99 [s, 9H, C(CH₃)₃]; HRMS (ESI) m/z calculated for $\text{C}_{27}\text{H}_{38}\text{O}_4\text{SSi}$ + Na: 509.2153, found: 509.2152. Analysis calculated for $\text{C}_{27}\text{H}_{38}\text{O}_4\text{SSi}$: C 66.62, H 7.87, S 6.59%; found: C 66.82, H 7.89, S 6.10%.

Crystallized from hexanes–EtOAc (m.p. 349–351 K). Data for (II): R_F 0.77 (5:1 hexanes/EtOAc); $[\alpha]_D^{25} +157.2$ (c 0.8, CH_2Cl_2); ^1H NMR (600 MHz, CDCl_3): δ 7.41–7.43 (m, 2H, ArH), 7.31–7.39 (m, 5H, ArH), 7.09–7.10 (m, 2H, ArH), 5.36 (d, 1H, $J = 5.4$ Hz, H-1), 4.80 (ABq, 2H, $J = 12.0$ Hz, PhCH₂), 4.32–4.34 (m, 1H, H-5), 4.13–4.15 (m, 1H, H-3), 3.97–3.98 (m, 1H, H-2), 3.95–3.96 (m, 1H, H-5'), 3.89–3.92 (m, 1H, H-4), 2.33 (s, 3H, ArCH₃), 1.08 [s, 9H, C(CH₃)₃], 0.99 [s, 9H, C(CH₃)₃]; HRMS (ESI) m/z calculated for $\text{C}_{27}\text{H}_{38}\text{O}_4\text{SSi}$ + Na: 509.2153, found: 509.2152. Analysis calculated for $\text{C}_{27}\text{H}_{38}\text{O}_4\text{SSi}$: C 66.62, H 7.87, S 6.59%; found: C 66.82, H 7.89, S 6.10%.

Crystal data

$\text{C}_{27}\text{H}_{38}\text{O}_4\text{SSi}$	$Z = 2$
$M_r = 486.72$	$D_x = 1.196 \text{ Mg m}^{-3}$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation
$a = 14.2256$ (15) Å	$\mu = 0.19 \text{ mm}^{-1}$
$b = 6.4531$ (7) Å	$T = 193$ (2) K
$c = 15.0426$ (16) Å	Prism, colorless
$\beta = 101.8651$ (14)°	$0.52 \times 0.41 \times 0.15 \text{ mm}$
$V = 1351.4$ (2) Å ³	

Data collection

Bruker SMART 1000 CCD	9410 measured reflections
PLATFORM area-detector	5524 independent reflections
diffractometer	4875 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.019$
Absorption correction: multi-scan	$\theta_{\text{max}} = 26.4^\circ$
(SADABS; Bruker, 2003)	
$T_{\text{min}} = 0.809$, $T_{\text{max}} = 0.972$	

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0466P)^2 + 0.1539P]$
$R[F^2 > 2\sigma(F^2)] = 0.036$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.091$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.22 \text{ e } \text{Å}^{-3}$
5524 reflections	$\Delta\rho_{\text{min}} = -0.15 \text{ e } \text{Å}^{-3}$
299 parameters	Absolute structure: Flack (1983),
H-atom parameters constrained	2491 Friedel pairs
	Flack parameter: 0.00 (6)

H atoms were placed in idealized positions (C–H = 0.95–1.00 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2003); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2003); software used to prepare material for publication: SHELXTL.

This work was supported by the Natural Sciences and Engineering Research Council of Canada, the Alberta Ingenuity Centre for Carbohydrate Science and the University of Alberta.

References

- Altona, C. & Sundaralingam, M. (1972). *J. Am. Chem. Soc.* **94**, 8205–8212.
 Bruker (2001). SMART (Version 5.054). Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2003). SADABS (Version 2.10), SAINT (Version 7.06A) and SHELXTL (Version 6.14). Bruker AXS Inc., Madison, Wisconsin, USA.
 D'Souza, F. W., Ayers, J. D., McCarren, P. R. & Lowary, T. L. (2000). *J. Am. Chem. Soc.* **122**, 1251–1260.
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.

- Han, J., Gadikota, R. R., McCarren, P. R. & Lowary, T. L. (2003). *Carbohydr. Res.* **338**, 581–588.
- Lemieux, R. U. & Koto, S. (1974). *Tetrahedron*, **30**, 1933–1944.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Yin, H., D'Souza, F. W. & Lowary, T. L. (2002). *J. Org. Chem.* **67**, 892–903.
- Yin, H. F. & Lowary, T. L. (2001). *Tetrahedron Lett.* **42**, 5829–5832.
- Zhu, X., Kawatkar, S., Rao, Y. & Boons, G.-J. (2006). *J. Am. Chem. Soc.* **128**, 11948–11957.